

j. How can the instantaneous rates of reaction

	Reactants	Products
Graph		
Instantaneous Rate Calculation		

k. Using a graph, show how the rate changes over time: (remember, the slope will give you the rate)

2. 6.1.2 Describe suitable experimental procedures for measuring rates of reactions. (2)

a. Reactions Producing a Gas

b. Change in Mass

c. Change in transmission or absorbance of light

d. Change in concentration measured with titration

e. Change in concentration measured with conductivity

f. Clock Reactions